

LITERATURE CITED

1. R. Shakirov, R. N. Nuriddinov, and S. Yu. Yunosov, *Khim. Prirodn. Soedin.*, 384 (1965).
2. A. Nabiev, R. Shakirov, and S. Yu. Yunusov, *Khim. Prirodn. Soedin.*, 535 (1975); 403, 676 (1976).
3. R. N. Nuriddinov, R. Shakirov, and S. Yu. Yunusov, *Khim. Prirodn. Soedin.*, 316 (1967).
4. R. N. Nuriddinov and S. Yu. Yunusov, *Khim. Prirodn. Soedin.*, 458, 767 (1971).
5. H. Budzikiewicz, *Tetrahedron*, **20**, 2267 (1964).
6. IUPAC-IUB Revised Tentative Rules for Nomenclature of Steroids, *J. Org. Chem.*, **34**, 1517 (1969).
7. R. N. Nuriddinov and S. Yu. Yunusov, *Khim. Prirodn. Soedin.*, 258 (1968).
8. R. F. Zurcher, *Helv. Chim. Acta*, **46**, 2054 (1963).
9. R. N. Nuriddinov and S. Yu. Yunusov, *Khim. Prirodn. Soedin.*, 260, 333, 334 (1968).
10. F. Bohlmann, *Ber.*, **91**, 2157 (1958).
11. T. M. Moynehan, K. Schofield, R. A. G. Jones, and A. R. Katritzky, *J. Chem. Soc.*, 2637 (1962).

THE STRUCTURE OF ARENINE

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From the phenolic fraction of the total alkaloids of *Papaver arenarium* M. B. we have isolated a new amorphous base $[\alpha]_D -31^\circ$ (c 1.03; CH₃OH) which we have called arenine (I) [1]. UV spectrum: λ_{\max} 244, 291, 319, 333 nm (log ϵ 4.49, 3.78, 3.57, 3.57). The IR spectrum shows absorption bands at (cm⁻¹) 930, 1040, (CH₂O₂), 1495, 1510 (aromatic ring), and 3300 (OH). In the mass spectrum there are the peak of the molecular ion with m/e 392 and also the peaks of ions with m/e 377, 363, 349, 135, and 84 (100%). The NMR spectrum of the base taken in deuteriochloroform showed the signals of three-proton singlets at 2.19 ppm from a N-methyl group and at 3.94 ppm from a methoxy group, and of two-proton singlets at 5.47 ppm from a methylenedioxy group and at 4.32 ppm from a methylene group. Six aromatic protons are represented by signals at 6.53-6.60 (3H), 7.47 (1H), 7.74 (1H), and 8.30 ppm (1H). Methylene and methine protons appear in the form of multiplets at 1.90-3.50 ppm.

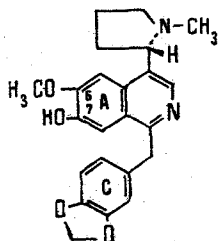
The UV, NMR, and mass spectra of arenine are close to the spectra of macrostomine [2]. On the basis of the facts obtained the following developed formula may be suggested for (I): C₂₀H₁₅N (N-CH₃) (CH₂O₂) (OCH₃) (OH).

A comparison of the developed formulas of macrostomine and arenine shows that the latter contains a hydroxy group in place of a methoxy group in macrostomine. When arenine was methylated with diazomethane, O-methylarenine (II) was obtained, which was identical according to TLC with macrostomine. A mixture of the hydrochlorides of (II) and of macrostomine gave no depression of the melting point. The hydroxy group in arenine may be present in ring A at C₆.

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or C₇. In the NMR spectrum of (I) the signal of the C₈ aromatic proton is shifted by 20 Hz (7.47 ppm) downfield as compared with that in the spectrum of macrostomine, and the protons of ring C (6.53, 6.60 ppm) and the methylene protons (4.32 ppm) are shifted upfield by approximately 10 Hz. These displacements are apparently connected with a change in the orientation of the benzyl part of the molecule of (I) relative to macrostomine and are due to the replacement of the methoxy group at C₇ in macrostomine by the hydroxy group in arenine.

The facts given above and the correlation of O-methylarenine with macrostomine show that arenine has the structure (I) with the S configuration of the asymmetric center



LITERATURE CITED

1. V. A. Mnatsakanyan, M. A. Manushakyan, and N. E. Mesropyan, *Khim. Prirodn. Soedin.*, 424 (1977).
2. V. A. Mnatsakanyan, V. Preininger, V. Simanek, J. Jurina, A. Klasek, L. Dolejs, and F. Santavy, *Collect. Czech. Chem. Commun.*, 42, 1421 (1977).

ALKALOIDS OF *Peganum nigellastrum*

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The main area of *Peganum nigellastrum* Bge. is the desert regions of Mongolia. The alkaloid composition of this plant has been little studied. Peganine and harmine have been found in the epigeal part by chromatography. We have investigated *Peganum nigellastrum* collected in the South Gobi iamak of Mongolia in the flowering phase. Chloroform extraction yielded 1.57% of combined alkaloids. The mixture of bases was separated into three fractions; A, B, and C. Fraction A was obtained by treating an acid extract with chloroform, fraction B consisted of the precipitate that deposited from the acid solution on alkalization, and fraction C was obtained by treating the filtrate with chloroform. Each fraction was separated further by preparative chromatography. Individual substances were crystallized or converted into salts.

Fraction A yielded a base in the form of a crystalline hydrochloride. From the latter a base was recovered with mp 110-111°C, mol. wt. 186; the R_f value of the substance on TLC [Al₂O₃; chloroform-benzene-methanol (5:4:1)] coincided with that of deoxyvasicinone [2], and a mixture with an authentic sample gave no depression of the melting point. The second base from this fraction was identified by TLC and a mixed melting point with an authentic sample of vasicinone [2].

From a base isolated from fraction B was obtained a nitrate with mp 166-168°C, which was identical with *dl*-peganine nitrate.

A base obtained from fraction C was identified as deoxyvasicinone.

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